

# Bioavailability of atrazine in char-amended soils

Yucheng Feng<sup>1</sup>, Vijay Loganathan<sup>1</sup>, Daniel  
Sheng<sup>2</sup>, and Prabhakar Clement<sup>1</sup>

<sup>1</sup>Auburn University &

<sup>2</sup>Zhejiang University of Technology

# Introduction

- Black carbon production:
  - Biomass burning: 50-260 teragrams C/year
  - Fossil fuel combustion: 12-24 teragrams C/year (Masiello & Druffel 1998)

- Biomass burning
  - Forests
  - Grasslands
  - Crop residues



Late-winter burn of mixed native grasses, Americus GA  
(Courtesy of Mary Goodman)

# Introduction

- Black carbon materials: charcoal, soot, char
- Black carbon contents in agricultural soils:
  - Australia: < 8 gC/kg soil, < 30% of SOC
  - U.S.: 1.8 - 13.6 gC/kg soil, < 35% of SOC

(Skjemstad et al. 2002)

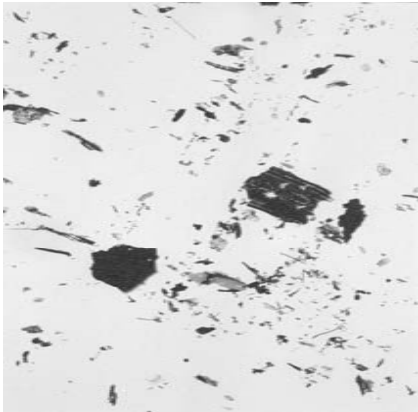
- Char → Sorption capacity ↑  
Bioavailability ?

# Objectives

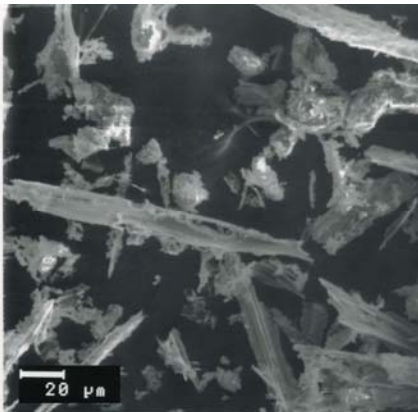
- Examine the impact of char on sorption, desorption, and bioavailability of atrazine in soils.

# Wheat char

Char image under light microscope (80x)



Scanning electron micrograph of char (500x)



- Prepared by burning in the field
- Characteristics:

	Wheat char	Activated carbon
Composition (wt%):		
C	12.9	87.6
Carbonate C	1.4	
Si	19.5	
K	20.9	
Ca	3.4	
Surface area (m <sup>2</sup> /g)	10.1	776

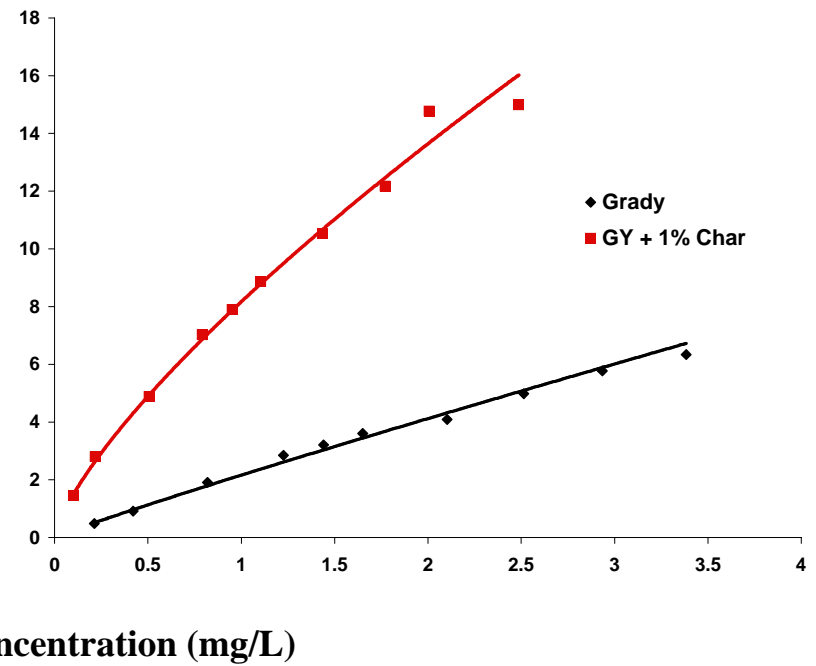
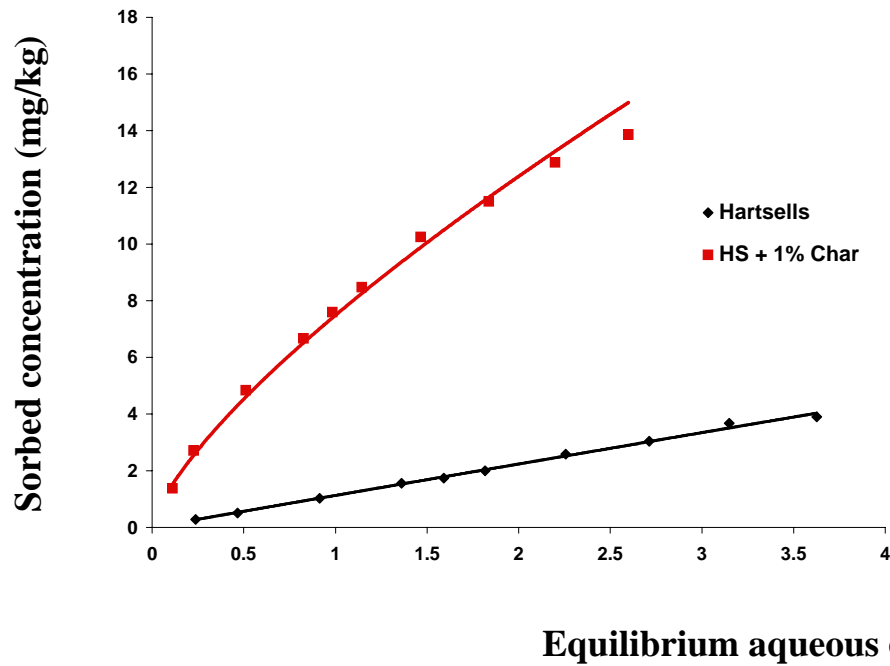
# Soils used in the study

	<b>Hartsells (HS)</b>	<b>Grady (GY)</b>
<b>Organic Carbon, %</b>	1.3	2.1
<b>pH</b>	5.3	5.7
<b>Texture</b>	Sandy loam	Clay loam
Sand, %	59.1	24.7
Silt , %	32.1	42.8
Clay, %	8.8	32.5
<b>Mineralogy</b>		
Kaolinite, % clay	60	45
Hydroxy-interlayered vermiculite, %	15	35
Quartz, %	15	10
Mica, %	5	-

# Methods

- Sorption
  - 1:10 soil to solution ratio
  - 1% char
  - 48-hr equilibration in phosphate buffer (pH 7)
- Desorption
  - Decantation and refill
- Biodegradation
  - Initial atrazine concentration: 2 mg/L
  - *Pseudomonas* sp. strain ADP
  - $^{14}\text{CO}_2$  production

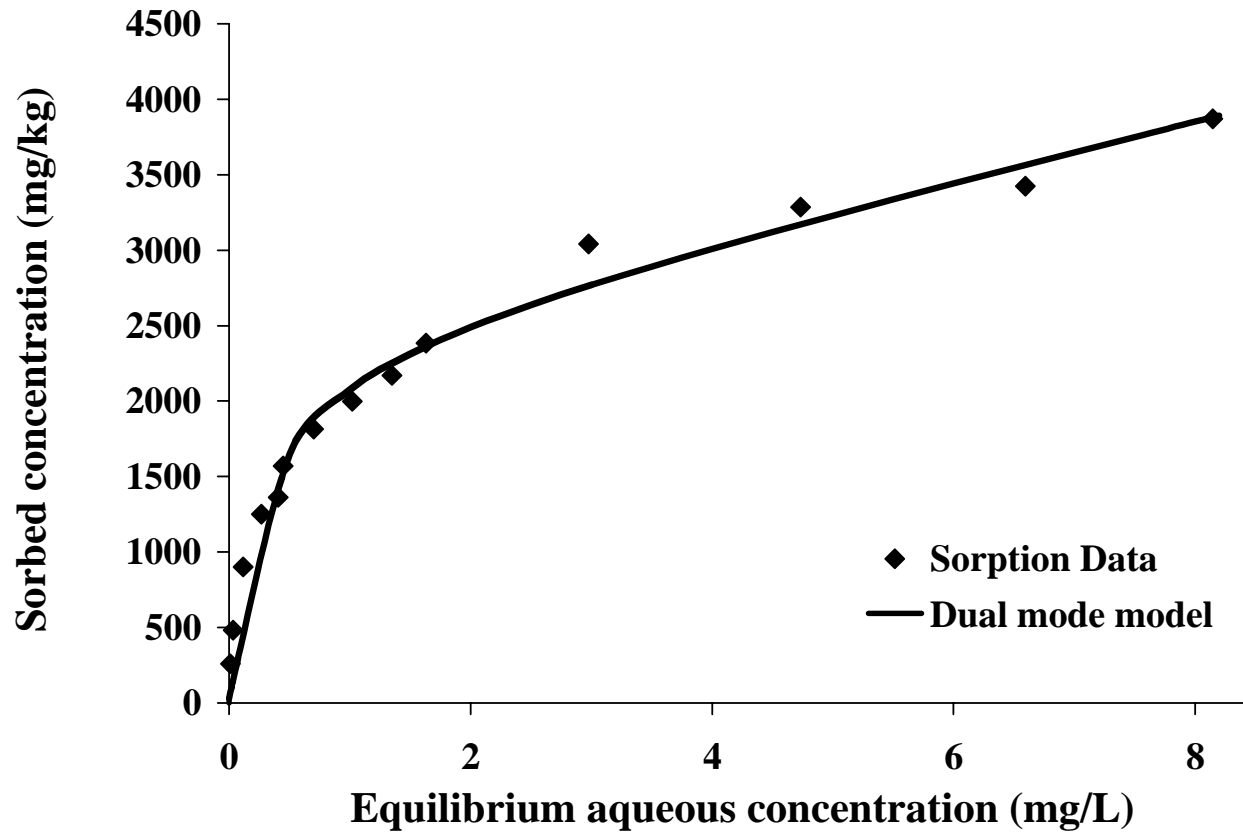
# Atrazine sorption by soils and char-amended soils



	$K_f$	$n$	$r^2$
Hartsells	1.49	0.96	0.997
HS+char	7.51	0.72	0.995

	$K_f$	$n$	$r^2$
Grady	1.92	0.88	0.990
GY+char	8.19	0.73	0.997

# Sorption of atrazine by char



$$q_e = \frac{Q_{\max} b C_e}{1 + b C_e} + K_p C_e$$

(Zhao et al. 2001)

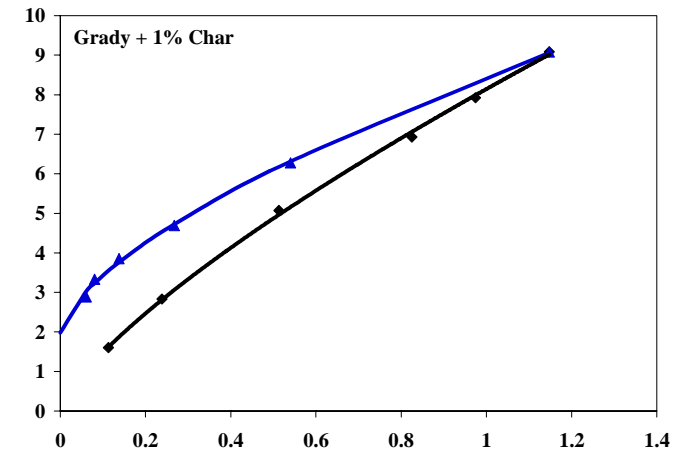
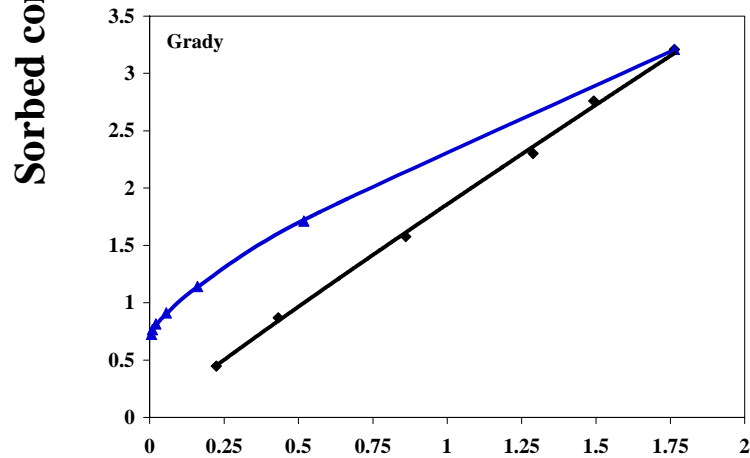
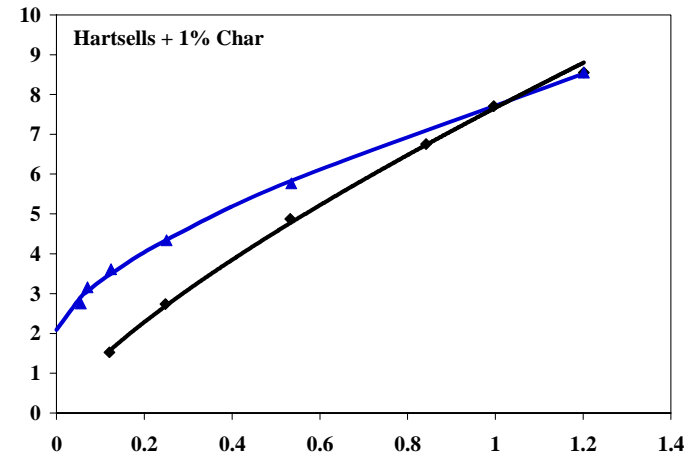
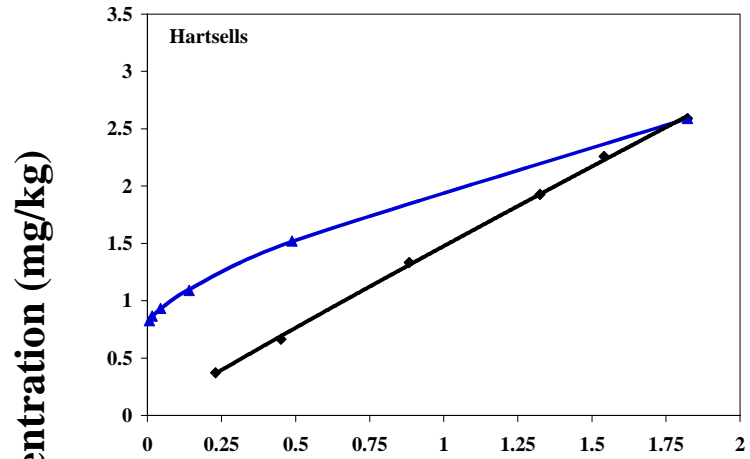
$$Q_{\max}: 2385 \text{ mg/Kg}$$

$$K_p: 194 \text{ L/Kg}$$

$$b: 3.8 \text{ L/mg}$$

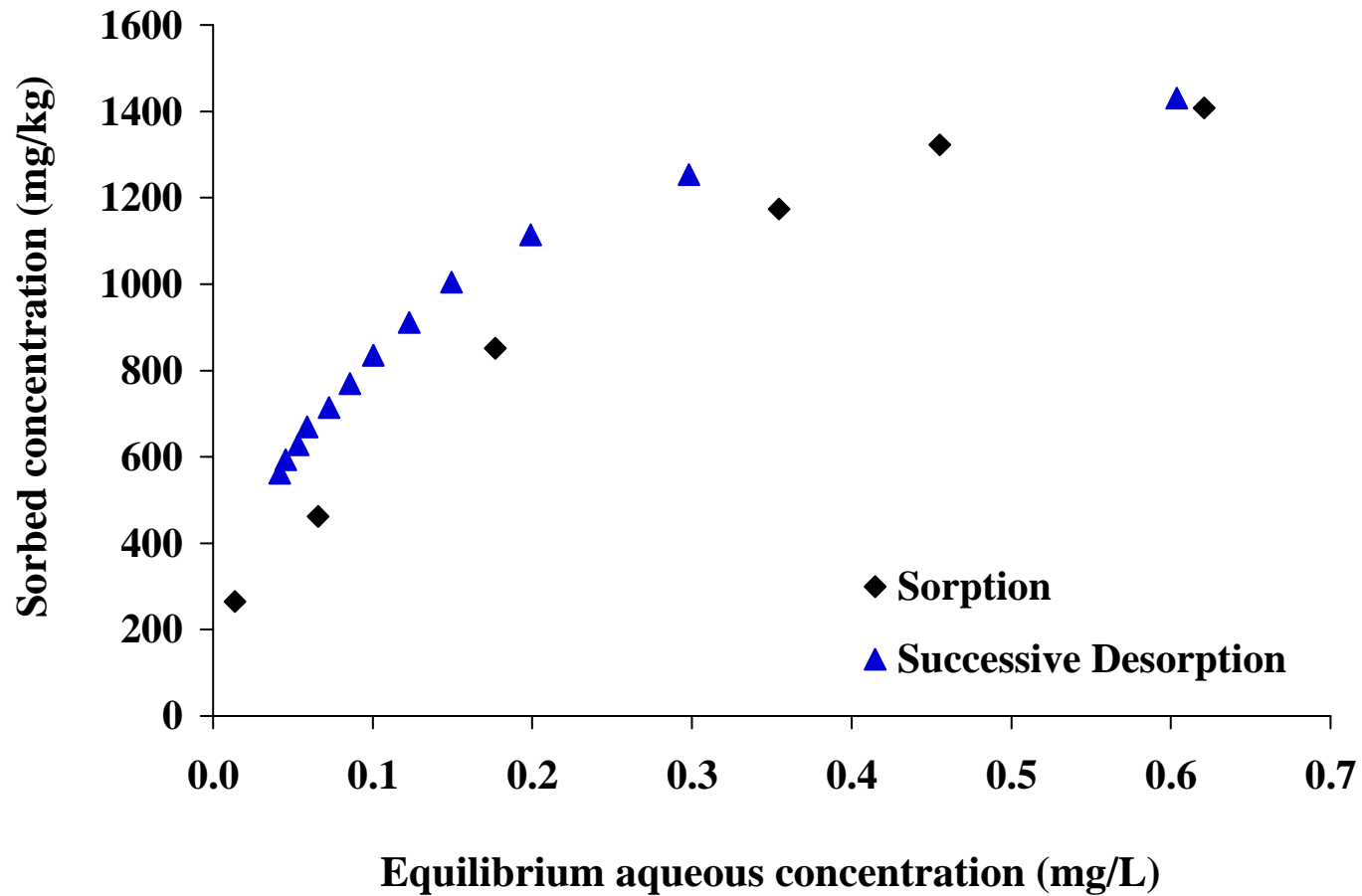
$$r^2 = 0.997$$

# Successive desorption of atrazine

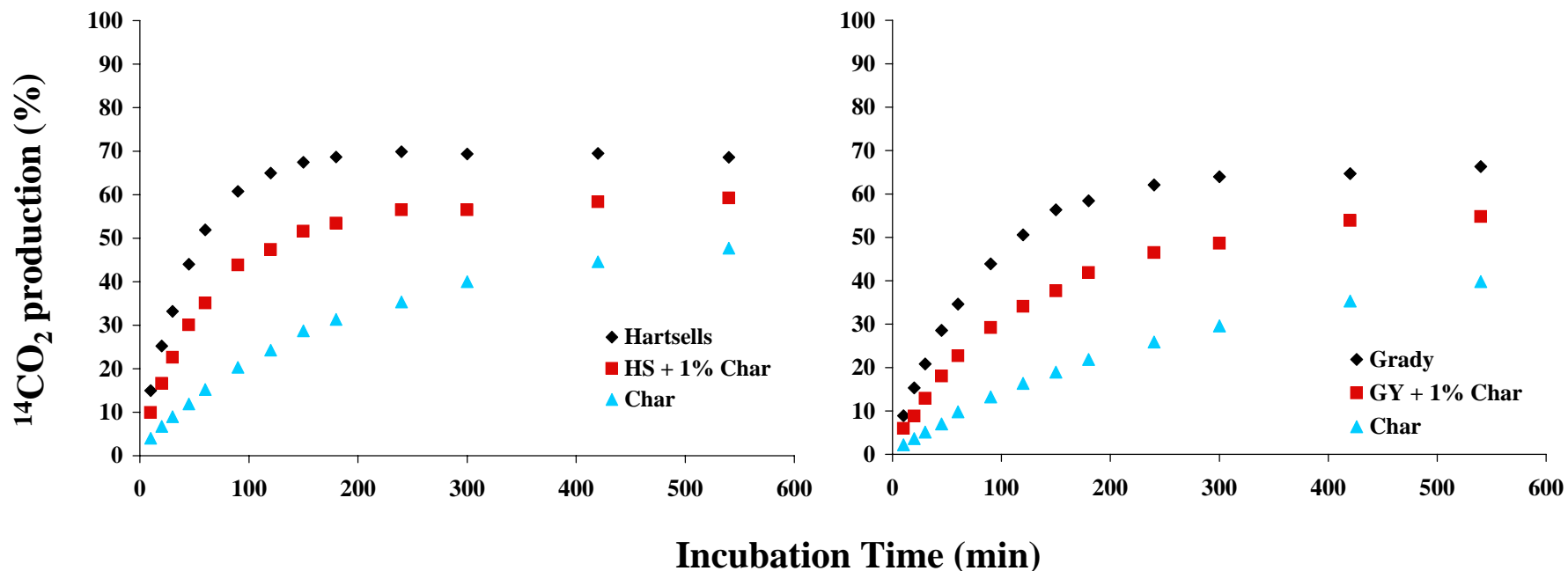


Equilibrium aqueous concentration (mg/L)

# Successive desorption of atrazine from char

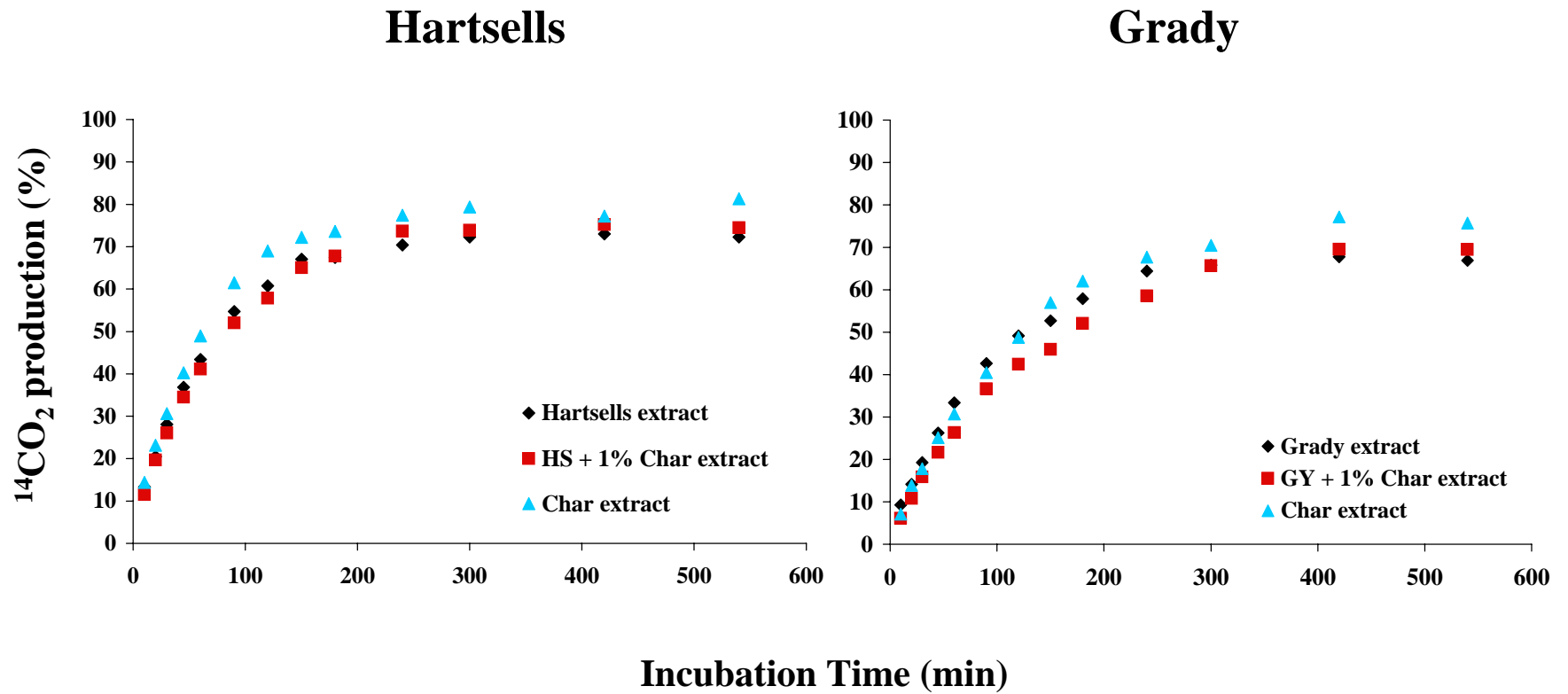


# Mineralization of atrazine by *Pseudomonas* sp. ADP



	Hartsells	HS+char	Char <sup>HS</sup>	Grady	GY+char	Char <sup>GY</sup>
P <sub>max</sub> , %	69.4	57.6	48.7	65.5	54.4	45.3
k (min <sup>-1</sup> )	0.0225	0.0159	0.0059	0.0126	0.0084	0.0037

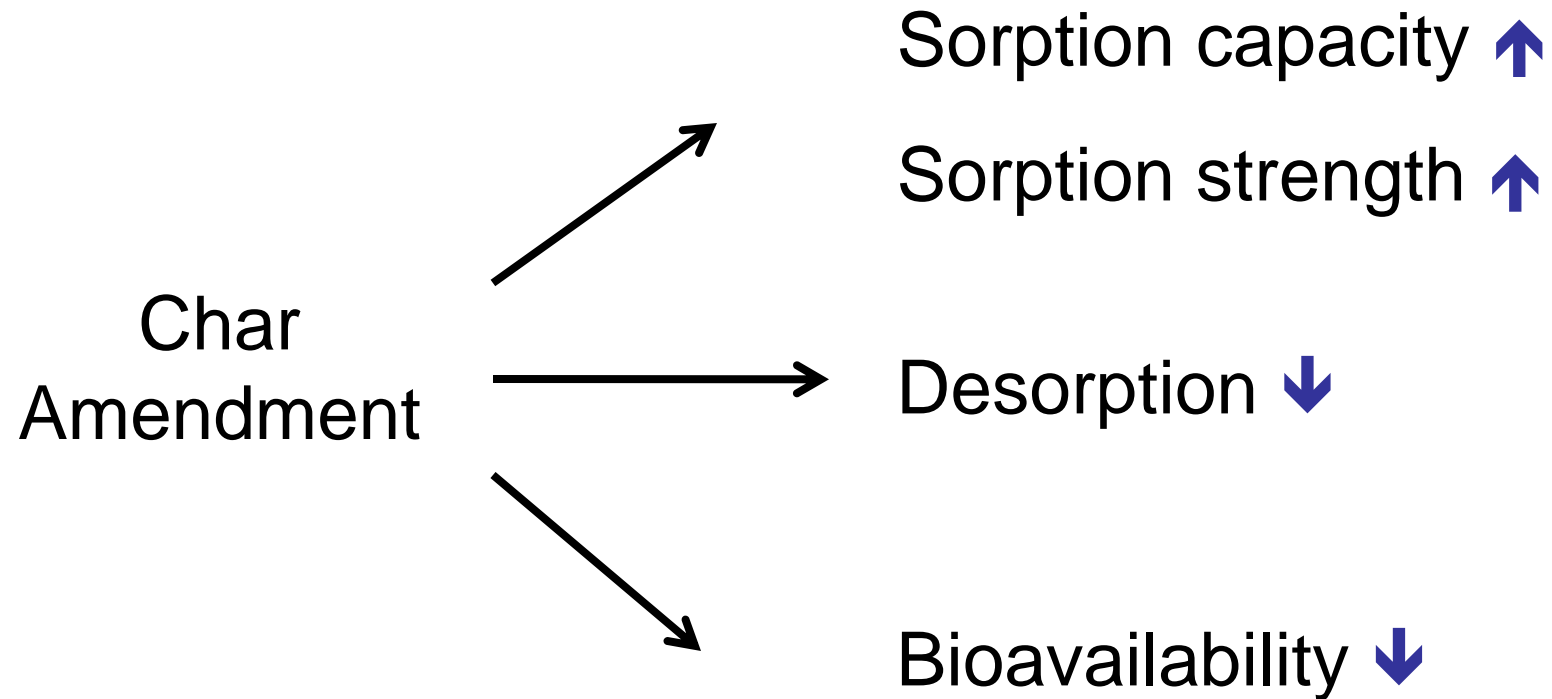
# Mineralization of atrazine in sorbent extracts



Initial  $^{14}\text{C}$  remaining (%) in sorbents after biodegradation, desorption, and extraction

	<b>Biodegradation</b>	<b>Desorption</b>	<b>Extraction</b>
<b>Hartsells</b>	$6.4 \pm 1.9$	$4.5 \pm 0.5$	$3.2 \pm 0.5$
<b>HS+char</b>	$28.1 \pm 2.2$	$21.9 \pm 0.2$	$5.3 \pm 0.1$
<b>Grady</b>	$5.1 \pm 0.3$	$5.3 \pm 0.1$	$3.5 \pm 0.04$
<b>GY+char</b>	$19.3 \pm 4.8$	$25.8 \pm 0.02$	$6.5 \pm 0.5$
<b>Char</b>	$10.4 \pm 8.1$	NA	NA

# Summary



# Acknowledgements

- USDA National Research Initiative
- Alabama Agricultural Experiment Station